

BBD-003-1104012

Seat No.

M. Sc. (CBCS) (Sem. IV) Examination

July - 2021

Chemistry: C(PA)-404

(Applied Analytical Chemistry) (New Course)

Faculty Code: 003

Subject Code: 1104012

Time : $2\frac{1}{2}$ Hours]

[Total Marks: 70

Instructions: (1) All questions carry equal marks.

(2) Attempt any five questions out of ten.

1 Answer the following:

- (a) Explain the term solvent extraction and give the type of extract system.
- (b) What are preservatives? Classify it with examples.
- (c) Give the function of blood.
- (d) Explain the term ore with example.
- (e) Give the chemical constituents of milk and mention the common adultrants of it.
- (f) Calculate the weight of Fe (III) left unextracted from 100 cm^3 of a solution having 200 mg of Fe³⁺ in 6 M HCl after three extraction with 25 cm^3 of ether. D = 150.
- (g) Give the parameters determining the green nature of analytical chemistry.

2 Answer the following :

- (a) What is process analytical technology?
- (b) How will you estimate sodium and potassium in blood sample?
- (c) Briefly explain microwave enhanced chemistry.
- (d) What is synergistic extraction? Give its mechanism.
- (e) What are food additives? Give the purpose of its addition.
- (f) How will you collect blood sample for glucose, CO₂ and HIV test from infected patients?
- (g) What is an alloy? Why alloys are used?

- 3 Answer the following:
 - (a) How will you analyze acid insoluble residue calcium oxide and magnesium oxide content in dolomite ore ?
 - (b) How will you analyze copper and zinc content in brass alloy?
- 4 Answer the following:
 - (a) How will you extract and analyze benzoic acid in the presence of saccharin in ready to serve beverages?
 - (b) Give the extraction, qualitative and titrimetric method for detection of benzoic acid present in beverages and liquid products.
- 5 Answer the following:
 - (a) What is the primary goal of PAT? How PAT works?
 - (b) Give the benefits of PAT for regulated industry.
- **6** Answer the following :
 - (a) Discuss techniques for solvent extraction in detail.
 - (b) How will you analyze serum calcium in blood sample?
- 7 Answer the following:
 - (a) Write note on green media used in analytical laboratories.
 - (b) Describe green methodology in analytical chemistry.
- **8** Answer the following :
 - (a) Describe the partition theory of solvent extraction.
 - (b) Explain solvent extraction by macromolecules in detail.
- **9** Answer the following :
 - (a) How will you analyze tin and lead content in solder alloy.
 - (b) How will you analyze sulphur-dioxide as preservative in food products?
- **10** Answer the following :
 - (a) How will you analyze chloride in blood sample?
 - (b) How will you estimate blood urea in blood sample?